TIP on HW tonight: #2) variables t & n are in days, not years. #3) Add 'annual' to question at end: What 'annual' interest rate did he charge you?						

## Applications of Logs

Compounded Interest:  $A = P \left( 1 + \frac{r}{n} \right)^{nt}$ 

Compounded Continuously: A = Pert

P = principal (amount invested)

r = rate as a decimal t = number of years

A = Amount after t years

n = number of compounds per year

e = Euler's number

Nov 10-9:00 AM

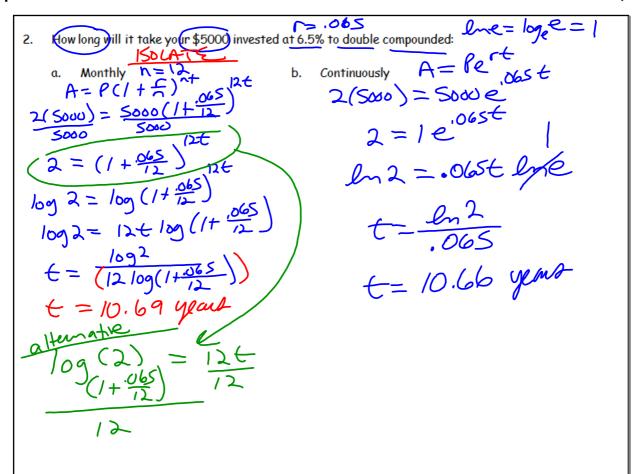
P A-B(1+=)~=.065

4=3.5

1. If \$5000 is invested at 6.5% annual interest rate, how much will you have after  $3\frac{1}{2}$  years if the money is compounded:

a. Annually 
$$n = (0.5)^{1(3.5)}$$
 $A = 5000(1 + \frac{1}{100})^{1(3.5)}$ 
 $A = 5000(1 + \frac{1}{100})^{1(3.5)}$ 

- b. Quarterly  $N = \frac{y}{A} = \frac{365}{4} = \frac{9(3.5)}{4}$  $A = \frac{56265.82}{4}$
- 12(3.5) d. Continuously  $A = Pe^{-\frac{1}{2}} .065(3.5)$ A = SODO = A = B(277.29



Nov 27-2:49 PM

3. If you put \$5000 in an account that pays interest quarterly what interest rate hust you receive in order to have \$7500 after 5 years?  $A = P(1 + F)^{n} + 4(5)$  7500 = 5000 (1 + F) 5000 = 5000 (1 + F) 5000 = 5000 (1 + F) 7500 = 7000 (1 + F) 7500 = 700

Mike invests \$6000 in a retirement account with a fixed annual interest rate compounded continuously) After 15 years, his balance is \$8099.15. What is the interest rate on the account?

$$\frac{8099.15}{6800} = e^{157}$$

$$\ln(\frac{8099.15}{6800}) = 157 \text{ fixe}$$

$$(= \frac{\ln(\frac{8099.15}{6800})}{15} = .01999 \approx 2.0\%$$

$$ln(\frac{8091.5}{6000}) = 15.501$$

Nov 10-9:03 AM

The mass of a radioactive element at time t is given by

$$A = A_0 \left(\frac{1}{2}\right)^{\frac{1}{h}}$$

Where Ao is the initial mass and h is the half-life of the element.

After 43 years, 20 milligram sample of strontium-90 (\*Sr) decays to 6.071 mg. What is the half-life of strontium-90?

$$\frac{6.071}{20} = (.5)$$

$$h \log(\frac{6.071}{20}) = (\frac{43}{h} \log(.5))h$$

$$h \log(\frac{6.071}{20}) = (43\log(.5))$$

$$h = \frac{43\log(.5)}{\log(\frac{6.071}{20})} = 25.00$$

$$25 \text{ years}$$

$$h = \frac{43\log(.5)}{\log(6.071)} = 25.00$$



When a living organism dies, its carbon-14 decays. The half-life of carbon-14 is 5730 years. If the skeleton of a mastodon has lost 58% of its original carbon-14 when did the mastodon die? (to the nearest hundred years)

die? (to the nearest hundred years)
$$A = A_{0} \left(\frac{1}{2}\right)^{4/4}$$

$$4\lambda = 1 \left(\frac{5}{2}\right)^{4/4}$$

100-58 1-.58 42% Emaring



$$t = \frac{5750 \log .42}{1113} = -6$$

$$log . 42 = \frac{1}{5730} log . 5$$
  
 $5730 log . 42 = tlog . 5$   
 $t = \frac{5730 log . 42}{log . 5} = 717/. 3 \text{ % 7200 yeard}.$ 

Nov 10-9:03 AM